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Section 2 – Project Information

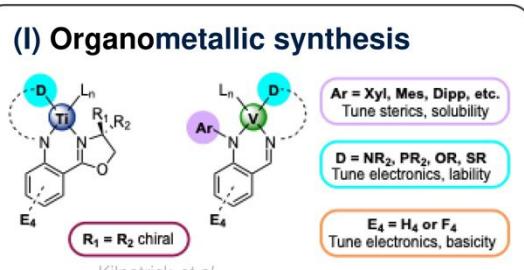
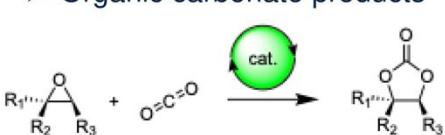
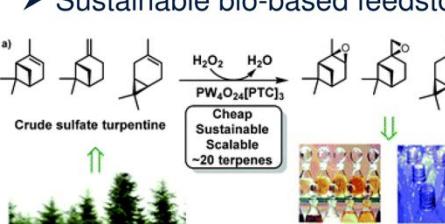
Project Title	Bio-Derived Carbonate Materials for Circular Chemical Manufacturing: Titanium Complexes for CO ₂ Valorisation
Project Summary	
CO ₂ chemistry and the development of reactions utilising CO ₂ as a C ₁ feedstock have drawn significant attention because CO ₂ is not only an abundant, inexpensive, and non-toxic carbon source but also a major contributor to climate change. One of the most promising reactions for using CO ₂ is its transformation with epoxides to yield cyclic carbonates (Figure 1). ^[1] Cyclic carbonates can be used as synthetic intermediates in the synthesis of fine or bulk chemicals. They have been used as a raw material for the synthesis of polycarbonates, and as components in other carbonate-containing materials and composites. Another application of cyclic carbonates, which has grown significantly in recent years, is as green polar aprotic solvents because of their excellent solubilising properties and relatively low toxicities. This has also led to their use as electrolyte solvents in lithium-ion batteries (LIBs). ^[2]	
The replacement of oil-derived oxiranes with a terpene-based epoxides, such as 1,2-limonene oxide, presents a pathway to new bio-derived carbonate materials. ^[3] This PhD project aims to develop a sustainable catalytic cycloaddition process, using CO ₂ and epoxides derived from terpenes, and study the material properties of the new carbonate products towards applications as green solvents.	
<div style="display: flex; justify-content: space-around;"> <div style="width: 45%;"> <p>(I) Organometallic synthesis</p>  <p><i>Kilpatrick et al.</i></p> </div> <div style="width: 45%;"> <p>(II) CO₂-epoxide cycloadditions</p> <ul style="list-style-type: none"> ➤ Organic carbonate products  ➤ Sustainable bio-based feedstocks  <p><i>Bull et al. Green Chem. 2021</i></p> </div> </div> <p>(III) Advanced applications</p> <p>LIB electrolytes</p>  <p>EV cooling liquids</p>  <p>Kumpan Electric, Bernd Dittich / Unsplash</p>	

Figure 1 Outline of the scientific objectives and technical skills to be gained from the project.

An approximate timeline and workplan for the 4 year PhD project is shown in Scheme 1.

Objective	Q1	Q2	Q3	Q4	Q5	Q6	Q7	Q8	Q9	Q10	Q11	Q12	Q13	Q14	Q15	Q16
I	Training + onboarding				Bimetallic $[Ti_2]$ synthesis & characterisation											
		Mono $[Ti]$ synthesis & characterisation					In situ studies via stop flow NMR, FT-IR, etc.								DFT calculations with collaborators	
II			CO ₂ -epoxide cycloaddition catalysts testing					Cycloadditions with terpene-derived epoxides								
III																Placement with industrial partner Thesis & publication write-up
																Material property testing for applications as solvents or coatings

Scheme 1 Timeline and objectives.

Kilpatrick *et al.* have a strong research focus on organometallic chemistry of the Earth abundant metals,^[4] including CO₂-epoxide copolymerisation,^[5] and have recently discovered a titanium-based catalyst that is highly active under mild conditions. Bull *et al.* have a wealth of experience developing sustainable routes to biorenewable feedstocks for applications in the green chemical industry.^[6] The project will be informed by engagement with industrial partners active in the development and deployment of functional fluids and electrolyte materials for energy storage and electric vehicle technologies.

This PhD project will expand the scope of early transition metal catalysts and bio-derived monomers applied to CO₂-epoxide cycloaddition reactions and address key questions on the technical and economic feasibility of carbonate products. The research will contribute to the development of more sustainable chemical processes and materials, supporting the transition towards a circular economy and the long-term goal of reducing net CO₂ emissions.

References

- [1] D. Prasad, K. N. Patil, N. K. Chaudhari, H. Kim, B. M. Nagaraja and A. H. Jadhav, *Catal. Rev. - Sci. Eng.*, 2022, 64, 356–443. DOI: 10.1080/01614940.2020.1812212.
- [2] V. Aomchad, À. Cristòfol, F. Della Monica, B. Limburg, V. D'Elia and A. W. Kleij, *Green Chem.*, 2021, 23, 1077–1113. DOI: 10.1039/D0GC03824E.
- [3] K. Xu, *Chem. Rev.*, 2004, 104, 4303–4417. DOI: 10.1021/cr030203g.
- [4] J. Allen, T. Krämer, L. G. Barnes, R. R. Hawker, K. Singh and A. F. R. Kilpatrick *Organometallics*, 2025, 44, 749. DOI: 10.1021/acs.organomet.4c00513.
- [5] A. C. Deacy, A. F. R. Kilpatrick, A. Regoutz and C. K. Williams, *Nat. Chem.*, 2020, **12**, 372–380. DOI: 10.1038/s41557-020-0450-3
- [6] J. D. Tibbetts, W. B. Cunningham, M. Vezzoli, P. Plucinski and S. D. Bull, *Green Chem.*, 2021, 23, 5449–5455. DOI: 10.1039/d1gc01734a.
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